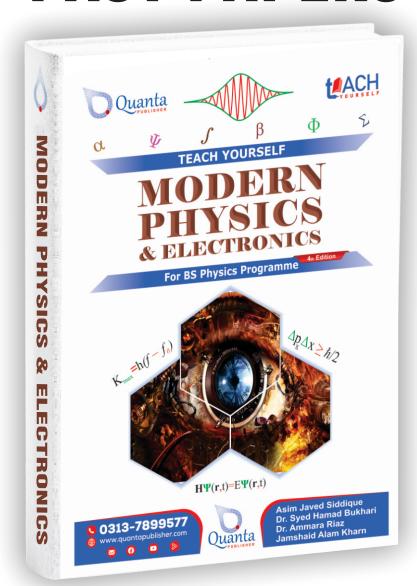
INTRODUCTION PAST PAPERS



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UNIVERSITY OF THE PUNJAB

Third Semester 2018
Examination: B.S. 4 Years Programme

		e: PHY-201/21330 TIME ALLOWED: 30 mins. MAX. MARKS: 10											
		Attempt this Paper on this Question Sheet only.											
		OBJECTIVE PART											
.1	Encircle the correct answer from given multiple choices in each part. (1 x 10)												
	A)	According to quantum physics, the energy of an atomic oscillator is											
		i) Constant ii) Continuous iii) Discrete iv) None of these											
	B)	What is the effect of the increasing the intensity of light in photoelectric effect											
		i) The KE of photoelectrons increases ii) The Stopping potential increases											
		iii) Cutoff frequency increases iv) Emission of photo electrons increases											
	C)	Series that lies in visible region of hydrogen spectrum is called											
		i) Lyman series ii) Ballmer series iii) Bracket series iv) Paschen series											
	D)	The magnitude of the minimum momentum of a particle, trapped in infinite potential											
		well of width L will be											
		i) 0 ii) $h^2/2L$ iii) $h^2/2L^2$ iv) $h/2L$											
	E)	Space quantization in magnetic field was experimentally verified by											
		i) Zeeman and Lyman ii) Einstein and de Hass iii) Stern and Gerlach											
		iv) Pauli and Bohr											
	F)	X-rays are produced through											
		i) Bremsstrahlung processes ii) K-shell emission processes											
		iii) Radioactive decay iv) Both Bremsstrahlung and K-shell emission											
	G)	The density of charge carriers in pure silicon at room temperature is of the order of											
		i) 10^{28} m ⁻³ ii) 10^{16} m ⁻³ iii) 10^{22} m ⁻³ iv) None of these											
	H)	When forward biased is increased across the PN diode, the depletion region											
		i) Increases ii) Decreases iii) Remains unchanged iv) None of these											
	I)	The activity or rate of decay of a radioactive source is measured in											
		i) Rad ii) Rem iii) Curie iv) Roentgen											
	J)	The emission of a beta particle from a nucleus results in											
		i) Decrease in the atomic mass number ii) Increase in the atomic mass number											
		iii) No change in the atomic mass number iv) None of these											

Third Semester 2018 Examination: B.S. 4 Years Programme Roll No.

PAPER: Concepts of Modern Physics Course Code: PHY-201/21330

TIME ALLOWED: 2 hrs. & 30 mins.

MAX. MARKS: 50

Attempt this Paper on Separate Answer Sheet provided.

SUBJECTIVE PART

Note: Attempt all questions of this section.

Q.2 Give the short answer of each question

 $(2 \times 10 = 20)$

- Explain thermal radiation.
- Explain the energy of oscillator by classical physics and quantum physics ii.
- iii. Briefly describe Wien's displacement law.
- Explain briefly the Einstein's photoelectric effect equation.
- v. Describe briefly, how Continuous X-ray spectrum is produced.
- vi. Why population inversion is necessary between two atomic levels for laser action to occur?
- vii. Explain semiconductors, according to band theory of solids.
- viii. Explain biasing rule for normal operation of NPN transistor.
- Describe the difficulties in operation of thermonuclear fusion reactor. ix.
- Give the difference between exothermic and endothermic reaction.
- Q.3 (a) Define de-Broglie's hypothesis. Explain in detail, how Davisson-Germer proved de Broglie's hypothesis. (1,6)
 - Calculate the de-Broglie's wavelength of a particle of mass 1.0 x 10⁻¹⁵ kg moving at (b)
- Q.4 State Bohr's postulates. Using Bohr's model of hydrogen atom, derive the expression (a) for total mechanical energy of electron orbiting about central proton.
 - (b) Calculate the binding energy of a hydrogen atom, that is, the energy that must be added to the atom to remove the electron from its lowest energy state. The values to be used; h=6.63 \times 10⁻³⁴ j.s, R=1.097 \times 10⁷ m⁻¹, C=3 \times 10⁸ m/s. (3)
- Q.5 What is fission chain reaction? Discuss the three problems together with their (a) solutions in working of nuclear reactor based on fission chain reaction. (1, 6)
 - Consider a 236U nucleus is in its ground state. How much energy is required to remove (b) a neutron from it, leaving a 235U nucleus behind? The needed atomic masses are 235U= 235.043924 u; n = 1.008665 u; 236 U= 236.045563 u. (3)



Fourth Semester - 2018 Examination: B.S. 4 Years

TIME ALLOWED: 15 Min.

Roll No.

PAPER: Physics-IV (Concepts of Modern Physics)
Course Code: PHY-213 / PHY-22307 Part - I (Compulsory)

MAX. MARKS: 10

Attempt this Paper on this Question Sheet only.

Please encircle the correct option. Each MCQ carries 1 Mark. This Paper will be collected back after expiry of time limit mentioned above.

Ouestion no: 1

(10x1=10)

Attempt all MCQs and chose the best answer.

- 1. Frequency below which no electrons are emitted from metal surface is
- a) minimum frequency
- b) angular frequency
- c) maximum frequency
- d) threshold frequency
- 2. Energy absorbed by electron is used in
- a) escaping the metal
- b) increasing kinetic energy
- c) both A and B
- d) increasing frequency
- 3. Microwaves have wavelength of about
- a) 10 cm
- b) 20 cm
- c) 30 cm
- d) 40 cm
- 4. Our eyes detect light in
- a) RGB form, Red Blue Green form
- b) ROYGBIV, rainbow color form
- c) The simple form of a particular color
- d) none of these ways
- 5. Moon is a good example of
- a) Luminous objects
- b) Non-luminous objects
- c) Transparent objects
- d) Opaque objects
- 6. Temperature of a gas is increased, its kinetic energy would
- a) increase
- b) decrease
- c) remain same
- d) increase and decrease both

- 7. Effect of diffraction is greatest if waves pass through a gap with width equal to
- a) frequency
- b) wavelength
- c) amplitude
- d) wavefront
- 8. Visible light has wavelength of
- a) 5 × 10⁻⁷ m
- b) $3 \times 10^8 \, \text{m}$
- c) 6 × 10³ m
- d) 4×10^4 m
- Spreading of wave as it passes through a gap or around an edge is called
- a) reflection
- b) refraction
- c) diffraction
- d) superposition
- In photoelectric effect, electrons should be removed from the
- a) inner shells
- b) surface
- c) from core
- d) the nucleus



Fourth Semester - 2018 Examination: B.S. 4 Years

PAPER: Physics-IV (Concepts of Modern Physics) Course Code: PHY-213 / PHY-22307 Part - II

Roll No. TIME ALLOWED: 2 Hrs. & 45 Min.

MAX. MARKS: 50

Attempt this Paper on Separate Answer Sheet provided.

Question no: 2

Write short answers of the following questions.

 $(4 \times 5 = 20)$

- 1. Define Planks distribution law and Wien displacement law?
- 2. Describe the working of Nuclear Reactor? Define Spectroscopy. Name three types of spectra?
- 4. Define azimuthal quantum number in detail?
- 5. State any 2 theorems associated with wave-functions in Quantum Mechanics?

Question no: 3

What do you know about Lorentz Transformation? Define Relativistic momentum in this regard?

Question no: 4

(10)

Describe the process of Controlled Thermo-Nuclear Fusion in detail?

Describe the theory of Beta-Decay in detail? Also calculate its penetration depth?

LO512	m: BS-4 Years in Chemistry Course Code: PHYS-206	Press Copy
O	am Name in which this course is common Bs Physics, Bs Mathematics / Chemistry.	
t-ourse	Filtle: Electronics and Modern Physics	5-247-1
Session	n; 2016 -2020 Semester 4th Final Time Allowed 02:00 hrs Max. Marks 50	
Q.No.	Questions	Marks
1.	Encircle the best option under each.	(8x1 =8)
	1. Which one of these is Accepter Atom?	
•	(a) B (b) P (c) Sb (d) As	
	2. A logic Gate in which an output is low only when both inputs are identical	
	(a) OR Gate (b) AND Gate (c) X-OR Gate (d) X -NOR Gate	19
	3. Exchange of energy Between radiations and material of a cavity is not continuous in	
		(
	(a) Wien's Law (b) Planck's Law (c) Stefan's Law (d) Rayleigh's Law	•
	4. If n=4 then the possible values of m _l are	Ç
	(a) 5 (b) 7 (c) 8 (d) 2	
	5. The relation for energy of Photon is	
	15 50 € bo/□	
	(a) $E = h \square / c$ (b) $E = h / f$ (c) $E = f / h$ (d) $E = h c / \square$	
	6. A transistor is used as	
	(a) Amplifier (b) Rectifier (c) Oscillator (d) Both a, and c	
	7. Which is called an inverte?	
	(a) OR Gate (b) AND Gate (c) NOT Gate (d) NANAD Gate	
	8 Wave -like nature of mater was verified by	
	(a)frank Hertz experiment (b) GP Thomson experiment (c) Stern Gerlach experiment	
	(d) Rutherford experiment	
	Write short answer of the following.	(2x10=20)
Q.2:	Write short answer of the tolowing.	
7	L Do pure semiconductors obey Ohm Law:	
	Do pure semiconductors obey Ohm ⁵ Law? When a diode is used as a rectifier?	
	2. Why a diode is used as a rectifier?	
	 Why a diode is used as a rectifier? Draw schematic symbol of X-NOR Gate. 	
	 2. Why a diode is used as a rectifier? Draw schematic symbol of X-NOR Gate. 4. What do you means by a negative feedback? 	
	 Why a diode is used as a rectifier? Draw schematic symbol of X-NOR Gate. What do you means by a negative feedback? State Stefan -boltz mann's Law 	
	 Why a diode is used as a rectifier? Draw schematic symbol of X-NOR Gate. What do you means by a negative feedback? State Stefan -boltz mann's Law Give postulates of Bohr's theory about Hydrogen Atom. 	
	 Why a diode is used as a rectifier? Draw schematic symbol of X-NOR Gate. What do you means by a negative feedback? State Stefan -boltz mann's Law Give postulates of Bohr's theory about Hydrogen Atom. How De Broglie interpreted about Bohr's orbit? Explain it. 	
	 Why a diode is used as a rectifier? Draw schematic symbol of X-NOR Gate. What do you means by a negative feedback? State Stefan -boltz mann's Law Give postulates of Bohr's theory about Hydrogen Atom. How De Broglie interpreted about Bohr's orbit? Explain it. Give an idea about Pauli Exclusion Principle. 	
	 Why a diode is used as a rectifier? Draw schematic symbol of X-NOR Gate. What do you means by a negative feedback? State Stefan -boltz mann's Law Give postulates of Bohr's theory about Hydrogen Atom. How De Broglie interpreted about Bohr's orbit? Explain it. Give an idea about Pauli Exclusion Principle. 	٤٠
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Q.3:	 Why a diode is used as a rectifier? Draw schematic symbol of X-NOR Gate. What do you means by a negative feedback? State Stefan -boltz mann's Law Give postulates of Bohr's theory about Hydrogen Atom. How De Broglie interpreted about Bohr's orbit? Explain it. Give an idea about Pauli Exclusion Principle. How Moseley Explained X -Ray Spectrum? Explain Briefly. 	
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Q.4:	 Why a diode is used as a rectifier? Draw schematic symbol of X-NOR Gate. What do you means by a negative feedback? State Stefan -boltz mann's Law Give postulates of Bohr's theory about Hydrogen Atom. How De Broglie interpreted about Bohr's orbit? Explain it. Give an idea about Pauli Exclusion Principle. How Moseley Explained X -Ray Spectrum? Explain Briefly. Derive Wien's Law from Planck's radiation Law LONG QUESTIONS (a) State De Broglie's Hypothesis. How can you verify this hypothesis by Davisson and Germer experiment? Explain it in detail. (b) In a Junction transistor, I_E = 5.82mA and I_B=120μA, calculate I_C (a) What do you mean by the Compton's effect? How can you Explain this effect on the basic of Quantitative analysis to derive an expression for Compton's shift? 	7+7+8=22
	 Why a diode is used as a rectifier? Draw schematic symbol of X-NOR Gate. What do you means by a negative feedback? State Stefan -boltz mann's Law Give postulates of Bohr's theory about Hydrogen Atom. How De Broglie interpreted about Bohr's orbit? Explain it. Give an idea about Pauli Exclusion Principle. How Moseley Explained X -Ray Spectrum? Explain Briefly. Derive Wien's Law from Planck's radiation Law LONG QUESTIONS (a) State De Broglie's Hypothesis. How can you verify this hypothesis by Davisson and Germer experiment? Explain it in detail. (b) In a Junction transistor, I_{II} = 5.82mA and I_B=120μA, calculate I_C (c) What do you mean by the Compton's effect? How can you Explain this effect on the basic of Quantitative analysis to derive an expression for 	7+7+8=22 5+2=7

Program: BS-4 Year in PHYSICS	_	Course	Code: PHYS	208
Course Title: <u>Electronics and Modern Physics</u>				
Session: 2017 - 2021 Semester 4TH	_Time Allowed	2 hrs	Max. Marks_	60

Q.No.	Questions	Marks							
Q.Nc.1	(a) Select the correct answer.	10							
	i. Semi conductor PN-junction is								
	(a) Rectifier (b) Transistor (c) Diode (d) None								
	ii. Diode which gives visible light when forward biased is called								
	i. Zener Diode (b) LCD (c) Photo diode (d) LED								
	iii. Power output in full wave rectifier than half wave rectifier is								
	(a) Same (b) Half (c) Double (d) None								
	iv. In negative feedback the phase difference between input and out put is								
	(a) 0° (b) 90° (c) 180° (d) 360°								
	v. if base current is very small then $l_c = \underline{\hspace{1cm}}$								
	(a) I _E (b) I _B (c) zero (d) infinite								
	vi. The current gain is given by								
	(a) , $\Delta I_E/\Delta I_B$ (b) $\Delta I_C/\Delta I_E$ (c) $\Delta I_C/\Delta I_B$ (d) None								
	vii. The electronic device that convert AC in DC is called.								
	(a) Oscillator (b) Amplifier (c) Rectifier (d) All								
	viii. Unit of RC is same as that of								
	(a) Energy (b) Time (c) Power (d) None								
	ix. There is a wide energy gap between valance and conduction band is								
	(a) semiconductor (b) conductor (c) insulator (d) metal								
	x. Panchen series of hydrogen spectra lies in region.								
	(a) Visible (b) IR (c) UV (d) far IR								
Q.No.2	Write the short answer to the following questions.	2×10=2							
	ii. Differentiate between insulator, semiconductor and insulator with the help of energy								
	band diagram. iii. What is the intrinsic semiconductor? iv. Why silicon is widely used in semiconductors?								
	v What is load line?								
	vi. Write the three configuration of a transistor with diagram. vii. What does light behaves as a particle and when does it wave?								
	viii. Can hydrogen atom be caused to emit X-ray? If so, describe how? If not, explain it.? ix. What will be the angle of scattering to get maximum Compton shift?								
	x. What is stopping potential in Photoelectric effect?								
Q.No.3	a) Describe the Davison and Germer experiment to verify de-Broglie hypothesis about the dual nature of radiation and matter.	08							
	b) Discuss Bohr's theory of postulates, find the value of Bohr's radius and the total energy of the								
	c) Explain the Photoelectric effect.	08							
	d) What are X-rays? Differentiate between k_{α} and k_{β} radiations.	08							
		06							
	BS-411-17-21-20.								

G.C.U.F. PAST PAPERS

		GOVERNMENT COLLEGE UNIVERSITY, FAISALABAD Affiliated Colleges Semester Examinations Roll No
BS Progr Course C Time Allo	ode:	PHY-403 Semester: III, BS Physics
		SUBJECTIVE
Attempt Al		-0/2
Q. No. 2:	(a)	Describe the Davison-Germer experimental evidence for the existence of de-Broglie wavelength with moving objects. (5)
	(b)	The wavelength of yellow spectral emission lines of sodium is 589 nm. At what
		kinetic energy would an electron have the same De-Broglie wavelength. (3)
Q. No. 3:	(a)	Describe LASER principle, working and properties of laser light (5)
	(b)	Calculate the ground state energy of electron in the hydrogen atom (2)
Q. No. 4:	(a)	Explain the phenomenon of Nuclear fission in detail (5)
	(b)	What are the units used for measuring ionizing radiation? (2)



B.S. 4 Years Program / Fourth Semester - 2019

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Paper: Physics-IV (Concepts of Modern Physics)
Course Code: PHY-213 / PHY-22307 Part – II

Time: 2 Hrs. 45 Min. Marks: 50

ATTEMPT THIS (SUBJECTIVE) ON THE SEPARATE ANSWER SHEET PROVIDED

- Q.2. Give short answers to the following.
- (4x5=20)
- 1. What is blackbody radiation? Write down Rayleigh's law and Planck's law for energy density and explain the difference between them. What is ultraviolet catastrphe?
- 2. The kinetic energies of photoelectrons range from zero to $4.0 \times 10^{-19} J$ when light of wavelength 3000 A falls on a surface. What is the stopping potential for this light?
- Determine the maximum scattering angle in a Compton experiment for which the scattered photon can produce a positron-electron pair.
- Calculate the kinetic energy of a neutron whose de Broglie wavelength is 0.7Å.
- Q.3. Give long answers to the following.

(3x10=30)

- (a) Find the energies of two photons that are produced when annihilation occurs between an electron and positron that are initially at rest.
 - (b) Dtermine a photon's threshold energy for pair production.

(5+5)

2. Discuss in detail the stability of nuclei.

(10)

3. State and explain de Broglie hypothesis. State properties of matter waves.

(10)

G.C.U.F. **PAST PAPERS**

GOVERMENT COLLEGE UNIVERSITY FAISALABAD



Discipline: BS- Physics

(4thSemester)

Time: 90 minutes.

Total Marks75(50+25)

Course Code: PHY-404 Credit Hours: 3(3-0)

Course Title: Modern Physics II Course Code
Final term Spring 2020)

0.0	Choose The Correct Answer		E0-1×E0		80=1×50		
г.							
0	Statement	A	В	С	3		
1	For S orbital l=	1	0	2	4		
2	Ellipse has axis	1	2	3	4		
3	The degenerate orbitals in p state	1	2	3			
4	When alpha particle is emitted ,atomic no decrease by	1	2	3	4		
5	Atomic no increase by,when beta particle is emitted	1	2	3	4		
6	Aomic no changes,when gama paticle is emitted	1	0	2	3		
7	Value of orbital quantum no l starts from	1	0	2	3		
8	The no of neutron in He4	2	1	5			
9	For d-shell l=2,the No .of Zeeman splits	3	5	7	(
10	For j=2, m ₁ have values	3	5	1	4		
11	Frank Hertz experiment was performed in	1918	1914	1920	1910		
12		1/2 IW ²	(1/2)Iw	Iw	None		
13	For p orbital I=1, S=1/2, j=	1/2,3/2	1/2,1/2	1/2,5/2	3/2,3/2		
14		10 ³ s	10 ⁻³ s	108s	10 ⁻⁸ s		
15	1 Mev=	105 ev	106 ev	10 ³ ev	10-6 ev		
16	atom in ground state	-13.6ev	13.6ev	10.4ev	1.4ev		
17	The magnetic quantum no can have	2(21+1)	2l+1	21-1	2(21-1)		
18		22.61ev	20.66ev	20.61ev	None of these		
19		2S+1	2S+2	2S+3	2S+4		
20	The 1st Excitation Potential of	4.9V	9.8V	10.4V	1.4V		
21		cicular	eliptical	linear	cylinderical		
22		CO	H ₂	CO ₂	All of these		
23	Hydrogen atom does not emit x-	Energy levels are too close	Energy levels are too apart	It is too small in size	It has single electron		
24		Ideal absorber	Ideal emitter	Both a&b	None of these		
25	X-ray Wavelength decreases if voltage	Increases	Decreases	Constant	None of these		
26		Increases	decreases	constant	None of these		
27		Infinite	zero	small	1,67x10-27 kg		
28		Ionic Bond	Covalent Bond	Polar Bond	None		
29	Nuclei with sameA but different Z	Isotopes	Isobars	Isitones	Isomers		
30		Joule-sec	Watt-sec	Newton-sec	None of these		
31	Unit of Intensity	m/watt	Watt/m ²	watt	None of these		
32	Laser Operation depend on	Meta state existence	Population Inversion	Both a&b	None of these		
33	Moment of Inertia	Mr ²	mr	M2r	2mr		

G.C.U.F. **PAST PAPERS**

34	Characteristics x ay depend on	Nature of target	Potential difference applied	Filament	None of thes
35	In He-Ne Laser action is produced by	Ne	Не	Both a&b	None
36	In Frank Hertz experiment ,Grid potential	Negative	positive	zero	None of These
37	Zeeman effect without spin	Normal Zeeman effect	Anomalous Zeeman effect	Photoelectric effect	Compton effect
38	The energy of bond electron	positive	Negative	Zero	None of these
39	Magnetic force	q(VxB)	B(qxV)	V(qxB)	none
40	Photon of high energy	Red colour	Green color	Violet colour	Blue colour
41	The state of the s	Sorbital	Porbital	D orbital	F orbital
42	If 2S+1=1	singlet	Doublet	Triplet	None of these
43	o - p - i - g - i o do c i d i i o i	Small Z	Large Z	Z	None of thses
44	The Stern Garlach Experiment confirms Quantization of electron	Spin	Angular Momentum	Energy	None
45	The Einstein coefficient A is related to rate of	Spontaneous emission	Stimulated emission	Continuous emission	None of these
46	According to Stefan Boltzman Law Intensity is proportional to	T ²	Т3	T ⁴	T ⁵
47	In Stern Garlach experiment Magnetic field	Uniform	Non uniform	No field	None of these
48	Bohr theory is unable explain	Zeeman effect	Multi electron	Fine structure	All of these
49	Splitting of spectral lines is greater in	Zeeman effect	Anomalous Zeeman effect	Stark effect	None of these
50	Splitting of spectral lines by electric field	Zeeman effect	Stark effect	Compton effect	None

Short Questions Marks:-25 Time:-50 minutes

Q.1.Define Orbital quantum Number.

Q.2.Differentiate between Characteristics X-rays and Continuous X-rays.

Q.3.Describe He- NeLaser.

Q.4. What is Ionic bond and Covalent, Bond?

Q.5.What is Zeeman effect?

<u>G.C.U.F.</u> **PAST PAPERS**

Govt. Postgraduate College of Science, Faisalabad.

Bs-Physics 3rd Semester M/E

Mid term examination

Course code:-phy -403

Time: - 50minutes

Course title:- Modern physics I

Marks:-08

(Subjective)

Note:-Attempt all question

Q.1.(a) State Wien displacement Law

Yazz.blogspot.com (b)Calculate the wavelength of maximum spectral radiency for the sun

With surface temperature of 5800k.

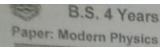
(3)

Q.2. What is Photo electric effect? Explain this effect using photon theory

And derive Enstein photo electric equation.

(4)

G.C.U.F. PAST PAPERS



B.S. 4 Years Program / Fourth Semester - Spring 2022

Course Code: PHYS-2002

ROLLN Time: 3 Hrs. Marks: 60

THE ANSWERS MUST BE ATTEMPTED ON THE ANSWER SHEET PROVIDED

Q.1. Briefly describe the following.

(15x2=30)

- . Write down the postulates of special theory of Relativity.
- ii. Define the term population inversion. Why population inversion is necessary between two atomic levels for laser action to occur?
- iii. What is q-value of nuclear rection? How does its positive or negative value effect the nuclear reaction?
- iv. Explain insulators, according to band theory of solids.
- Find the energy released during the alpha decay of 238U. The needed atomic masses are 38U= 238.050785 u; 234Th= 234.043593 u; 4He= 4.002603 u;
- what is relativistic momentum? Give its relation.
- vii. Define cooper pair.
- viii. What is inverse beta decay?
- ix. Mention four characteristics of laser.
- X. Differentiate between boson and fermions.
- Xi. Write down the general nuclear reaction showing how atomic and mass number of an atom is affected with the emission of gamma radiation.
- xii. Differentiate between absorption and emission spectrum with the aid of diagram.
- xiii. Write down the Maxwell Boltzmann distribution function.
- Lxiv. Why does the classical theory fail to explain specific heat of solids?
 - xv. What is the purpose of moderator in nuclear reactor?

Answer the following questions

(3x10=30)

- Q2. Write down Planck's law for energy density. Why is it superior to Rayleigh's law?
- Q3. State Bohr's postulates. Using Bohr's model of hydrogen atom, derive the expression for total mechanical energy of electron orbiting about central proton. Show that the electronic energy states are discrete
- Q4. What are semiconductors? How N-type/P-type semiconductor is formed? Describe the formation of depletion region in a junction diode? Can we control the width of depletion